

**Responsible Care Performance Reporting – Performance Indicators Questionnaire 2017:
Definitions and Further Explanations**

APPENDIX I

CORE PARAMETERS

1) Number of Fatalities (employees only)

- **Definition:** A fatality is an instantaneous work-related event or exposure, leading to death within one year.
- The parameter is expressed as **number of fatalities**.

Example:

In Company A in 2002, there was one incident resulted in the death of 2 employees in 2002. The “Number of Fatality for Employees” is therefore 2 for Company A in the year 2002.

2) Lost Time Injuries Frequency Rate (employees only)

- **Definition:** A lost time injury is an instantaneous bodily defect so that the individual is physically or mentally unable – as determined by a competent medical person – to work on a scheduled day or shift, resulting in at least 3 days off the job.
- The frequency rate is expressed as the **number of lost time injuries per million working hours**.
= $\frac{a \times b}{c}$ where a = total number of accidents
b = 1,000,000 man-hours
c = total man-hours worked
- Employee is defined to include all permanent full or part-time employees.
- Contractor is defined as any non-company employee who is providing a service to the company on the company’s premises. This definition does not differentiate between a contractor’s direct employees and those of any sub-contractors.

3) Sulphur Dioxide (SO_x)

- **Definition:** Airborne emission of sulphur and its compounds formed during combustion, production or treatment processes.
- The parameter comprises SO₂ and SO₃ and is expressed as **tonnes SO₂**.

Formula for Method 1:

Sulphur Oxides can be calculated from the amount of Fuel Oil consumed in a year and the average concentration of Sulphur (S) in the Fuel Oil.

$$SO_2 \text{ (Tons/year)} = \frac{\text{Fuel Oil used (tons/year)} \times \text{Concentration of Sulphur in fuel oil (\%)} \times 64}{32}$$

where the molecular weight of SO₂ = 64 and that of Sulphur = 32

Example:

The annual total consumption of Fuel Oil in Company A is 1250 tons, and the average concentration of Sulphur (S) in the Fuel Oil is 0.015%. Calculate the Sulphur Oxides released.

Amount of Fuel Oil (tons) = 1250 tons
% of Sulphur in Fuel Oil = 0.015%
Molecular Weight of Sulphur Oxides, MW_{SO2} = 64
Molecular Weight of Sulphur, MWs = 32
Sulphur Oxides released (tons) = 1250 (tons) x (0.015/100) x (64/32) = 0.375 (tons)

Formula for Method 2:

Sulphur Oxides can be calculated from the analysis result of Sulphur Oxides Concentration (either SO₂ or SO₃) in the stack gas, and the known flowrate of the stack gas.

$$\text{Sulphur Oxides (Kg/hour)} = \frac{(\text{Conc SO}_2 \text{ (ppm)} \times \text{Mol wt}_{\text{SO}_2} \times \text{Stack gas Flowrate (M}^3\text{/hr)})}{22.4 \times 10^6}$$

$$\text{Sulphur Oxides (Tons/year)} = \frac{(\text{Sulphur Oxides (kg/hour)} \times 24 \text{ hours} \times \text{No of Working days/year})}{10^3}$$

Example:

Average SO₂ concentration in a boiler stack gas is 1004 ppm. The stack gas flowrate is 4,467 m³/hour. Calculate the emission of SO_x from the stack?

$$\begin{aligned} \text{SO}_2 \text{ (kg/hour)} &= 1004 \times 64 \times 4467 / (22.4 \times 10^6) &= & 12.81 \text{ kg/hour} \\ \text{SO}_2 \text{ (Tons/year)} &= 12.81 \times 24 \times 300 \text{ days/year} / 1000 &= & 92 \text{ tons/year} \end{aligned}$$

4) **Nitrogen Oxides (NO_x)**

- **Definition:** Airborne emission of compounds of nitrogen and oxygen from combustion, production or treatment processes.
- The parameter comprises NO and NO₂ and is expressed as **tonnes of NO₂**.

Formula for Method 1:

Oxides of Nitrogen (NO_x) can be calculated from the analysis result of NO_x Concentration in the stack gas, and the known flowrate of the stack gas.

$$\text{NO}_x \text{ (Kg/hour)} = \frac{(\text{Conc NO}_x \text{ (ppm)} \times \text{Mol wt}_{\text{NO}_x} \times \text{Stack gas Flowrate (M}^3\text{/hr)})}{22.4 \times 10^6}$$

$$\text{NO}_x \text{ (Tons/year)} = \frac{(\text{NO}_x \text{ (kg/hour)} \times \text{Annual Operating Hours})}{10^3}$$

Example:

Average NO_x concentration in a boiler stack gas is 200.6 ppm. The stack gas flowrate is 4,491 m³/hour and the annual operating hours of the boiler was estimated as 7200 hours. Calculate the emission of NO_x from the stack?

$$\begin{aligned} \text{NO}_x \text{ (kg/hour)} &= 200 \times 38 \times 4491 / (22.4 \times 10^6) &= & 1.52 \text{ kg/hour} \\ \text{NO}_x \text{ (tons/year)} &= 1.52 \times 7200 \text{ hours/year} / 1000 &= & 10.9 \text{ tons/year} \end{aligned}$$

Where molecular weight of NO_x = 38, NO = 30 and NO₂ is 46.

Convert NO to NO₂:

For NO_x emissions, the reporting requirement is to be expressed as NO₂, if your facility measuring NO it has to be divided with 0.6522 to convert to NO₂.

Example:

The concentration of NO and NO₂ in a stack gas is 50 and 200 ppm respectively. The stack gas flowrate is 4,000 m³/hour. There was a shutdown in the continuous operating plant for two months. Calculate the emission of NO_x from the stack?

$$\begin{aligned} \text{NO}_x \text{ (kg/hour)} &= [(50/0.6522) + 200] \times 46 \times 4000 / (22.4 \times 10^6) &= & 2.27 \text{ kg/hour} \\ \text{NO}_x \text{ (Tons/year)} &= 2.27 \times 24 \text{ hours/day} \times 30 \text{ days/month} \times 10/1000 &= & 16.3 \text{ tons/year} \end{aligned}$$

5) Chemical Oxygen Demand (COD)

- **Definition:** Chemical Oxygen Demand (COD) is the amount of oxygen required for the chemical oxidation of compounds in water, as determined using a strong oxidant (most standard methods use dichromate).
- The parameter is expressed as **tonnes of oxygen**.
- For sites that have their wastewater treated at a shared third party unit and cannot obtain individual data, the efficiency factor of the wastewater treatment unit should be taken into consideration when calculating the amount.

Formula:

Chemical oxygen demand can be calculated from the average concentration of the COD and the annual discharge of the effluent to outfall. We can also estimate total COD released to environment based on the carbon content in the waste.

Method #1 (COD emitted from the actual analytical data):

$$\text{COD emitted (Tons/year)} = (\text{Ave COD conc. (mg/L} \times \text{Total Liquid waste (tons/year)} \times 10^{-6}$$

Average COD concentration can be calculated by taking the average of the daily reading or take the average of random sample results during the year. In case of TOC results multiply the TOC results with ~3 (32/12) to estimate COD.

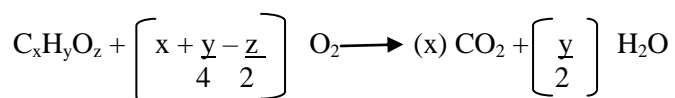
If the facility has more than one outfall need to calculate for different streams and add up for calculating total COD discharged during the reporting period.

$$\text{Total COD emitted (Tons/year)} = \text{COD (tons/year)}_{\text{stream1}} + \text{COD (tons/year)}_{\text{stream2}}$$

Method #2 (Theoretical estimation):

The COD can be calculated from theoretically by knowing the waste characteristics. Below is a general equation for estimating COD from an organic waste.

Where $C_xH_yO_z$ is the molecular formula of the material and x, y, z in the number of atoms of



Carbon, Hydrogen and Oxygen present in one molecule.

Example:

The average COD concentration of the effluent discharged in the site outfall for the year was 85 ppm and the daily total outfall flow rate was 5 tons. Calculate the total COD discharged to environment?

$$\begin{aligned} \text{Daily Average Wastewater flow rate} &= 5 \text{ tons/day} \\ \text{COD concentration} &= 85 \text{ ppm} \\ \text{Quantity of COD in tons/year} &= 5 \text{ tons/day} \times 85/1000000 \times 365 \text{ days/year} \end{aligned}$$

6) Energy Consumption

- **Definition:** Energy consumption is the amount of fossil fuels and electricity (the result of purchased electricity plus self-produced **renewable** electricity minus electricity sold to the network). The unit for electricity is the GWh but in order to add them up, electricity should be converted to TOE.

- The theoretical conversion factors are: 2.3876×10^{-5} ktoe per GJ and 0.2332 ktoe per GWh.
- The parameter on energy consumption is expressed as **tonnes of fuel oil equivalent (TOE)**.

Formula:

$$\text{Energy Consumption} = \text{Energy Content of Fuel(s)} + \text{Electricity Consumed}$$

To calculate Energy Content of Fuel(s), consider:

- i) Energy Content of a fuel is the product of multiplication the Amount of Fuel Consumed (tons) and its Heat Value found in Table 1.

Example: 50 tons for Residual Fuel Oil, its Heat Value in Table 1 is 42 GJ/ton.

The Energy Content is therefore $50 \text{ (ton)} \times 42 \text{ (GJ/ton)} = 2100 \text{ (GJ)}$

To convert to the right unit from GJ to toe, is to multiply 2.3876×10^{-5} . The Energy Content in this example is $2100 \text{ (GJ)} \times 2.3876 \times 10^{-5} = 50.1207 \text{ (toe)}$.

- ii) If two or more types of fuel are consumed, the total Energy Content is the summation of energy contents of their respective fuels consumed.

To calculate Electricity Consumed, consider:

- i) Electricity Consumed is the net result of purchased electricity from the Electricity Generating Company minus electricity sold by the Reporting Company.

The unit for electricity consumed may be in kWh, MWh or GWh.

(1 GWh = 1,000 MWh = 1,000,000 kWh). For reporting purpose, Electricity Consumption should be converted to tons of fuel oil equivalent (toe).

Conversion: for simplicity, the theoretical conversion factors to be used are 2.3876×10^{-5} ktoe per GJ and 0.2332 ktoe per GWh.

Example:

A Reporting Company consumed 1000 tons of fuel oil and 1,500 tons of natural gas per year to generate steam and small amount of electricity. In addition, it consumed 3,500 GWh of electricity purchased from the grid and sold 500 GWh of electricity to a nearby Company. Note that any electricity produced by the Reporting Company should be accounted for under fuel consumed

From Table 1 – Typical Heating Values of various fuel types

- Heating value of fuel oil = 42 GJ/ton
- Heating value of natural gas = 51 GJ/ton

Energy Content of Fuel(s) consumed per year
 $= 1,000 \text{ tons} \times 42 \text{ GJ/ton} + 1,500 \text{ tons} \times 51 \text{ GJ/ton}$
 $= 118,500 \text{ GJ}$ Convert GJ to toe

Energy Content of Fuel(s) consumed per year
 $= 118,500 \text{ GJ} \times 2.3876 \times 10^{-5}$
 $= 2.829 \text{ ktoe} = \underline{2829 \text{ toe}}$

Electricity Consumed per year
 $= \text{Purchased Electricity} - \text{Exported Electricity}$
 $= (3500 \text{ GWh} - 500 \text{ GWh})$
 $= 3000 \text{ GWh}$

(Convert to toe) Electricity Consumed per year
 = 3000 GWh x 0.2332
 = 699.6 ktoe
 = 699,600 toe

Energy Consumption of the Reporting Company for the year
 = Energy Content of Fuel(s) + Electricity Consumed
 = 2829 toe + 699,600 toe
 = 702,429 toe

7) **Carbon Dioxide (CO₂)**

- **Definition:** The major contribution to CO₂-emissions by the chemical industry is from the combustion of fuels both directly and indirectly associated with the usage of electricity. Therefore, these emissions are calculated on the basis of energy consumption.
- **Direct Carbon Monoxide (CO₂):** The direct emissions of CO₂ are calculated as tons of CO₂ equivalent by multiplying the amount of solid, liquid and gaseous fuels used for energy use and for the generation of self produced electricity, by corresponding CO₂-emission factors (see Annex I).
- **Indirect Carbon Dioxide (CO₂):** The indirect emissions of CO₂ are calculated as the multiplication of the amount of net purchased electricity by the average factor of CO₂ emissions per kwh produced.

Formula:

Carbon Dioxide = Direct Carbon Dioxide + Indirect Carbon Dioxide

- Direct Carbon Dioxide (CO₂)
 Direct CO₂ is emissions caused by the Reporting Company either burning fuel to generate electricity, steam or other purposes. Direct CO₂ emissions are calculated based on the sum total of all fuels consumed by the Reporting Company.
- Indirect Carbon Dioxide (CO₂)
 Indirect CO₂ is emissions of CO₂ from 3rd parties that supply electricity of steam to the Reporting Company. As a result of fuels combustion by the 3rd parties, CO₂ emissions are generated. Indirect CO₂ consists of the sum of net electricity purchased and net steam purchased.

Net electricity purchased is the net of electricity purchased from the grid (Electricity Supply Company) minus electricity sold to the grid/others. Net steam purchased is the net of steam purchased from 3rd parties minus steam sold to another 3rd parties.

Example:

A Reporting Company consumed 1,000 tons of fuel oil and 1,500 tons of Natural Gas per year to generate its own steam and small amount electricity. In a year it consumed 3,500 GWh of electricity purchased from the grid and sold 500 GWh of electricity to a nearby Company. In addition, the Company imported 10,000 tons of steam in the year and exported 500 tons of the steam purchased to the nearby company. Note that any electricity produced by the Reporting Company should be accounted for under fuel consumed.

- Calculating Direct Carbon Dioxide

For Fuel Oil:

From Table 1, Heating value of Fuel Oil = 42 GJ/ton
 Therefore, Energy Content of Fuel Oil = 1000 tons x 42 GJ/ton = 42,000 GJ

From Table 2, CO₂ Emission Factor for Fuel Oil is 77.4 (Kg CO₂ per GJ)

Therefore, CO₂ emission from Fuel Oil Combustion = 42,000 GJ x 77.4 Kg/GJ = 3,250,800 Kg

For Natural Gas:

From Table 1, Heating value of Natural Gas = 51 GJ/ton
Therefore, Energy Content of Fuel Oil = 1500 tons x 51 GJ/ton = 76,500 GJ

From Table 2, CO₂ Emission Factor for Natural Gas is 56.2 (Kg CO₂ per GJ)
Therefore, CO₂ emission from Natural Gas Combustion = 76,500 GJ x 56.2 Kg/GJ = 4,299,300 Kg

Total Direct Carbon Dioxide = 3,250,800 Kg + 4,299,300 Kg = 7,550,100 Kg = 7550.1 tons of CO₂

- Calculating Indirect Carbon Dioxide

a) For Electricity:

From Table 3, the International Electric Grid Factor for CO₂ is 0.575 tons CO₂ per MWh or 575 tons per GWh.

Net Electricity purchased = 3,500 GWh – 500 GWh = 3,000 GWh

Therefore, CO₂ emission due to Electricity purchased
= 3,000 GWh x 575 ton/GWh
= 1,725,000 tons of CO₂

b) For Steam:

Assume the steam is purchased from a 3rd party source firing natural gas

From Table 4, CO₂ Emission Factor for purchased steam = 0.14 tons CO₂ per ton Steam.

Net Steam purchased = 10,000 tons – 500 tons = 9,500 tons

Therefore, CO₂ emissions due to Steam purchased
= 9,500 tons x 0.14 = 1330 tons CO₂

Total Indirect Carbon Dioxides = 1,725,000 tons + 1,330 tons = 1,726.330 tons

Carbon Dioxide = Direct Carbon Dioxide + Indirect Carbon Dioxide
= 7542.5 tons + 1,726,330 tons = 1,733,872.5 tons CO₂

8) Other Greenhouse Gases

- **Definition:** Other Greenhouse Gases, listed in the Kyoto Protocol are:-

- (i) Methane (CH₄)
- (ii) Nitrous Oxide (N₂O)
- (iii) Hydrofluorocarbons (HFC's)
- (iv) Perfluorocarbons (PFC's)
- (v) Sulphur Hexafluoride (SF₆)

- Emission data for these gases should be reported **as tons of CO₂ equivalents**. The impact of the release of these gases on Climate Change is calculated by multiplying the tons release per year by its Global Warming Potential (GWP) relative to carbon dioxide, as published by the Intergovernmental Panel on Climate Change (IPCC).

Formula:

Other Greenhouse Gas = Tons of the Greenhouse Gas Emission x Global Warming Potential of the Greenhouse Gas.

Example:

A Reporting Company estimates a 15 tons of Methane (CH₄) emission last year.

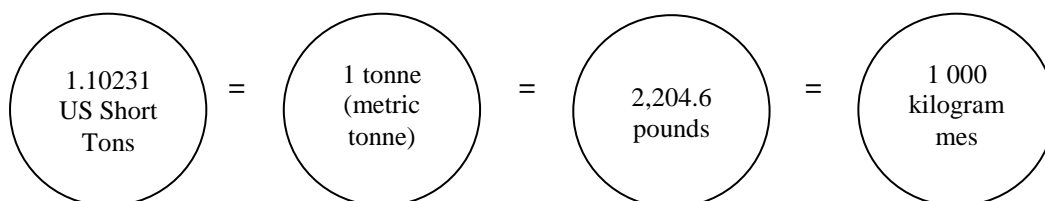
From Table 5 – GWP for Methane (CH₄) = 23 equivalent to CO₂

Therefore, the Other Greenhouse Gases – Methane (CH₄) emission in that year = 15 tons x 23 equivalent to CO₂ = 345 tons CO₂ equivalent

9) **Other Explanation**

- **Tonne:** A tonne is a Metric Tonne equivalent to 1000 kg and equivalent to 1.10231 US Short tons weighing 2,204.6 pounds.

1 Tonne = 1,000 kilograms = 2,204.6 pounds = 1.10231 US Short Tons



- **Volume transported:** Tonnes of chemical product shipped via air, rail, road, sea, inland waterway or pipeline between the site of a supplier company and that of the final customer. Includes the transport and off-site loading/unloading at ports, airports, warehouses, etc. and excludes the transport and loading/unloading activities at the premises of the supplier chemical company and the final customer.
- **Conversion Factors and Formulas**

TABLE 1: Typical Heating Values for Various Common Fuels

| Fuel Type | Heating Value (GJ/ton fuel) |
|--|-----------------------------|
| Natural Gas | 51 |
| Gasoline | 47 |
| Distillate | 45 |
| Residual Fuel Oil | 42 |
| LPG | 50 |
| Typical Fuel Gas generated from a Refinery or Chemical Plant | 50 |

TABLE 2: Carbon Dioxide Emissions Factors

| Fuel | Carbon Emission Factors ¹ (kg C/GJ) | Carbon Dioxide Emission Factor ¹ (kg CO ₂ /GJ) | Carbon Dioxide Emission Factor ¹ (tonne CO ₂ /toe) | Carbon Dioxide Emission Factor ² (lb CO ₂ /MMBTU) |
|-------------------|---|---|---|--|
| Crude Oil | 20.0 | 73.4 | 3.1 | 160.60 |
| Gasoline | 18.9 | 69.4 | 2.9 | 151.77 |
| Kerosene | 19.6 | 71.9 | 3.0 | 157.39 |
| Jet Fuel | 19.5 | 71.6 | 3.0 | 156.59 |
| Motor Gasoline | 20.2 | 74.1 | 3.1 | 162.21 |
| Residual Fuel Oil | 21.1 | 77.4 | 3.2 | 169.43 |

| Fuel | Carbon Emission Factors¹ (kg C/GJ) | Carbon Dioxide Emission Factor¹ (kg CO₂/GJ) | Carbon Dioxide Emission Factor¹ (tonne CO₂/toe) | Carbon Dioxide Emission Factor² (lb CO₂/MMBTU) |
|---------------------|--|--|--|---|
| Naphtha | 20.0 | 73.4 | 3.1 | 160.60 |
| Bitumen | 22.0 | 80.7 | 3.4 | 176.66 |
| Lubricants | 20.0 | 73.4 | 3.1 | 160.60 |
| Refinery Feedstocks | 20.0 | 73.4 | 3.1 | 160.60 |
| Other Oil | 20.0 | 73.4 | 3.1 | 160.60 |
| Steam Coal | 25.8 | 94.7 | 4.0 | 207.17 |
| Coking Coal | 25.8 | 94.7 | 4.0 | 207.17 |
| Petroleum Coke | 27.5 | 100.9 | 4.2 | 220.83 |
| Lignite | 26.1 | 95.8 | 4.0 | 209.58 |
| Sub-bituminous Coal | 27.6 | 101.3 | 4.2 | 221.63 |
| Peat | 28.9 | 106.1 | 4.4 | 232.07 |
| BKB & Patent Fuel | 25.8 | 94.7 | 4.0 | 207.17 |
| Coke | 29.5 | 108.3 | 4.5 | 236.89 |
| Natural Gas (dry) | 15.3 | 56.2 | 2.4 | 122.86 |
| Natural Gas Liquids | 15.2 | 55.8 | 2.3 | 122.06 |
| LPG | 17.2 | 63.1 | 2.6 | 138.12 |

Sources:

1. Greenhouse Gas Inventory Workbook Volume 2; IPCC/OECD Joint Programme; see conversion example below.
2. Environmental Protection Agency, inventory of US Greenhouse Gas Emissions and sinks: 1990-2006, April 15, 2006; referenced from Intergovernmental Panel as Climate Change (IPCC), Second Assessment Report (SAR).

Examples for Calculating CO₂ Emissions:

| | Carbon Emission Factor (kg C/GJ) | Molecular Weight Ratio of CO₂/C (44/12=3.67) | CO₂ Emission Factor (kg CO₂/GJ) | CO₂ Emission Factor (tonne CO₂/toe) |
|-------------|---|--|--|--|
| Steam Coal | 25.80 | 3.67 | 94.69 | 3.961 |
| Crude Oil | 20.00 | 3.67 | 73.33 | 3.070 |
| Natural Gas | 15.30 | 3.67 | 56.15 | 2.349 |

Note: 1 toe = 41.868GJ

TABLE 3: International Electric Grid Emission Factors

| Country | CO₂ | CH₄ | N₂O | Total (CO₂ Equivalent) |
|----------------|-----------------------|-----------------------|-----------------------|--|
| Malaysia | 0.575 | 2.67E-06 | 8.64E-06 | 0.578 |

TABLE 4: Carbon Dioxide Emission Factors for Purchased Steam

| Fuel Type | Carbon Dioxide Emission Factor for Purchased Steam (tons CO₂ / ton steam) |
|------------------|---|
| Natural Gas | 0.14 |
| Fuel Oil | 0.209 |

TABLE 5: Global Warming Potential (GWP) of Other Greenhouse Gases (GHG)

| GHG | GWP (based on the effects of greenhouse gases over 100-year time horizon) |
|------------------|--|
| CO ₂ | 1 |
| CH ₂ | 23 |
| N ₂ O | 296 |
| HFC _s | From 97 to 12,000 – depending on the HFC |
| PFC _s | From 5,700 to 11,900 – depending on the PFC |
| SF ₆ | 22,200 |

DIAGRAM 1: Electricity to Tonnes of Fuel Oil Equivalents

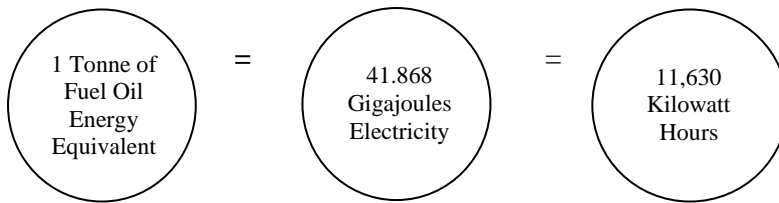
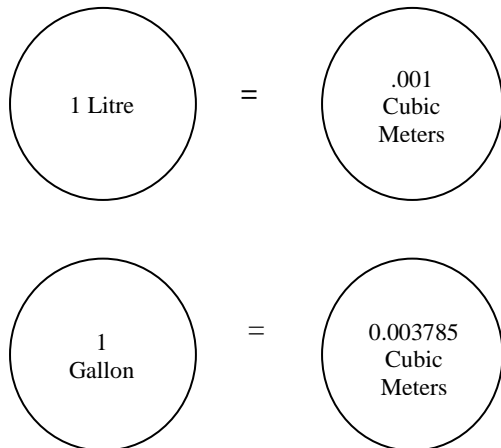


DIAGRAM 2: Water Consumption – Conversions to Cubic Meters



APPENDIX II

Transport Incidents

The distribution indicator relates to all incidents involving the movement of hazardous and non-hazardous chemicals. It covers the modes of transport by road and rail and includes the following:-

- death or injury
- spill or leakage
- property damage and/or material loss
- no spill or leak but public disruption
- adverse media attention

Incidents are categorized according to whether they were severe or less severe and whether distribution was by bulk or packaged cargoes.

| Definition of Distribution Incidents Showing Breakdown by Severity | | |
|--|---|---|
| <i>Type of Incident</i> | <i>Severe</i> | <i>Less Severe</i> |
| Death / Injury | Incident resulting in more than 3 days off from work | Incidents resulting in 1 – 3 days off work or treatment at approved medical centre |
| Spillage / Leakage | Hazardous material leakage of 200 kg or more Non-hazardous material leakage of 1000 kg or more | Hazardous material leakage of 10 kg or more but less than 200 kg. Non-hazardous material leakage of 100 kg or more but less than 1000 kg |
| Property Damage / Material - Loss including cost of environmental clean-up | In excess of RM50,000 | Between RM5,000 and RM50,000 |
| Public Disruption / Perceived Danger - Resulting in evacuation, road closure or other precautionary measures including attendance of emergency services | Last more than one hour | Last less than one hour |
| Adverse Media Coverage | National Level | Local Level |

APPENDIX III

Total Worker Hours

RCLG Associations should report the total number of employee hours worked for each member company in their association **and** the total number of contractor hours worked for each member company as a **combined, single number**. For the purposes of this guidance document, each association should refer to their local and regional definitions for employee and contractor. The goal for reporting total hours is to include **all individuals** who are involved with chemical manufacturing, except where those individuals are tasked with major construction projects such as large scale investments with specific, one-time project organizations created for design, engineering, and construction of new or significant expansion to existing process facilities. When reporting total worker hours, companies should report the same hours used for reporting **personnel hours**. This way, companies can have the same data set for occupational and process safety. Personnel hours should include those hours from **all** chemical manufacturing operations, not just from facilities where a process safety incident occurred.

Process Safety Event

For the purposes of this ICCA Reporting, a process safety event has occurred when:

- A. When a chemical substance or a chemical process is directly involved; **AND**
- B. The incident occurred in production, distribution, storage, utility, pilot plant within the site boundaries of company's facility; **AND**
- C. There was a release of material or energy (e.g. fire, explosion, implosion) from a process unit; **AND**
- D. One or more of the following **Reporting Thresholds** have been met:
 1. **Safety / Injury**
 - Injury resulting in a Recordable, Lost Time Accident or Fatality; or Hospital admission of anyone on or off site; **OR**
 2. **Direct Damage Cost**
 - A fire, explosion or clean up necessary to avoid/remediate environmental damage resulting in a direct cost equal to or greater than \$2,500 USDs; **OR**
 3. **Shelter in Place / Evacuation**
 - An officially declared shelter in place (on or off site); **OR**
 - An officially declared evacuation (on or off site); **OR**
 - A precautionary off site shelter in place or evacuation **OR**
 4. **Threshold Release**
 - The material released meets one of the GHS thresholds in **Table 1 – GHS Classification Table** (*measured in amount released during one hour*)

| Health Hazards | | 1 | 2 | 3 | 4 | 5 |
|--|-------------------|---------------------------------|--------------------|---|---|----------------|
| Acute Toxic (GHS 3.1) category: | | | | | | |
| Germ Cell Mutagenicity, Carcinogenicity, Reproductive toxicity, STOT-single exposure (GHS 3.5 – 8) | | | | | | All categories |
| All other health hazards (GHS 3.x) | | | | | | All categories |
| Physical Hazards (GHS 2.x) | | | | | | All categories |
| Environmental Hazards (GHS 4.x) | | | | | | All categories |
| Equivalent classification using GHS Hazard Statements: | H300, 310, or 330 | H301, 302, 311, 312, 331 or 332 | Any other H-Number | | | |
| Release Thresholds (During 1 hour timeframe) | ≥ 1 kg | ≥ 10 kg | ≥ 100 kg | | | |